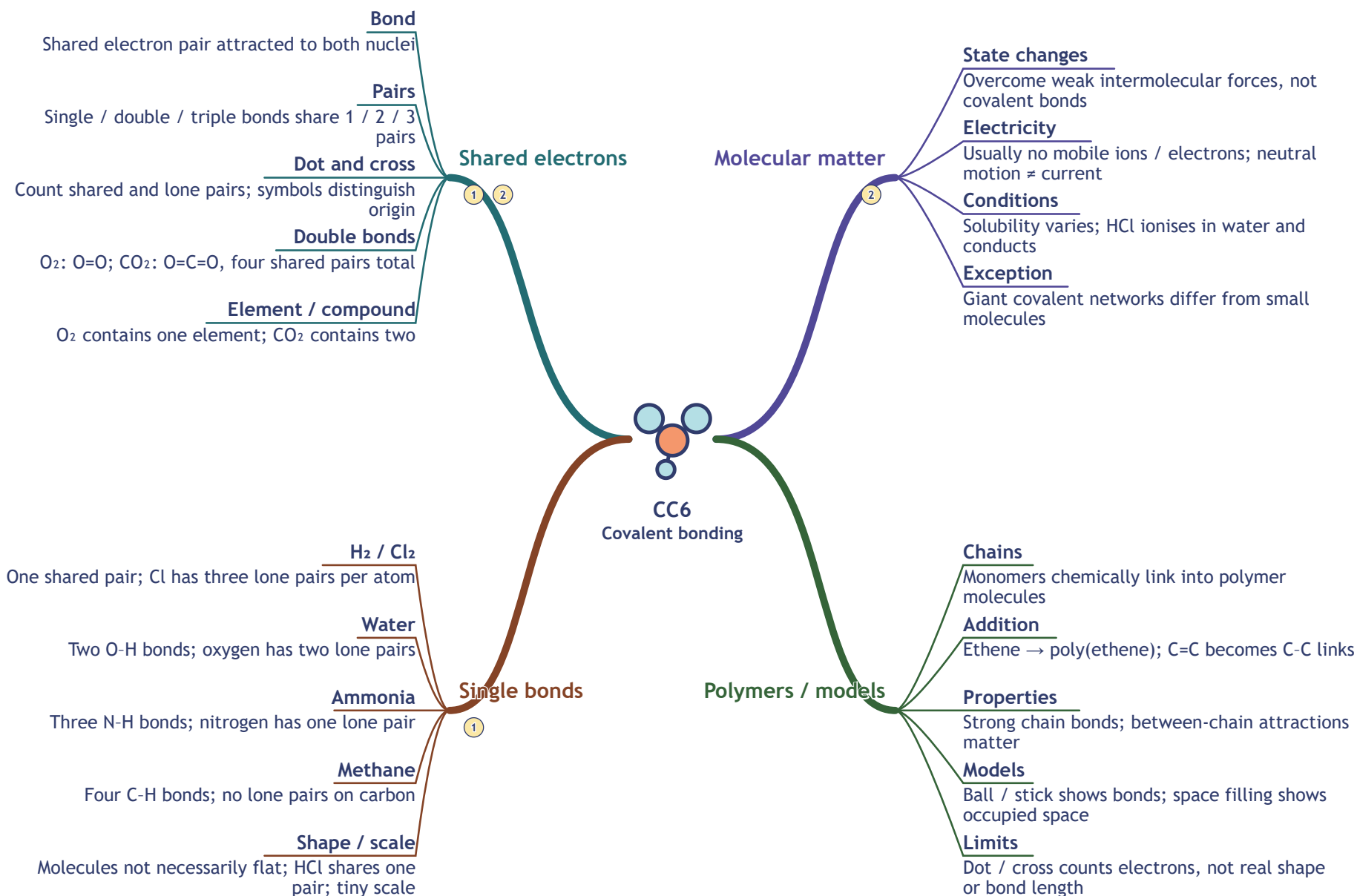




1. Shared electrons → Single bonds: Shared pairs build the bonds in molecules.

2. Shared electrons → Molecular matter: Strong bonds within molecules differ from forces between them.